

Novel One-Pot Synthesis of 4-Trifluoromethyl-2,4-dienamides

Yanchang Shen* and Yuejun Xiang

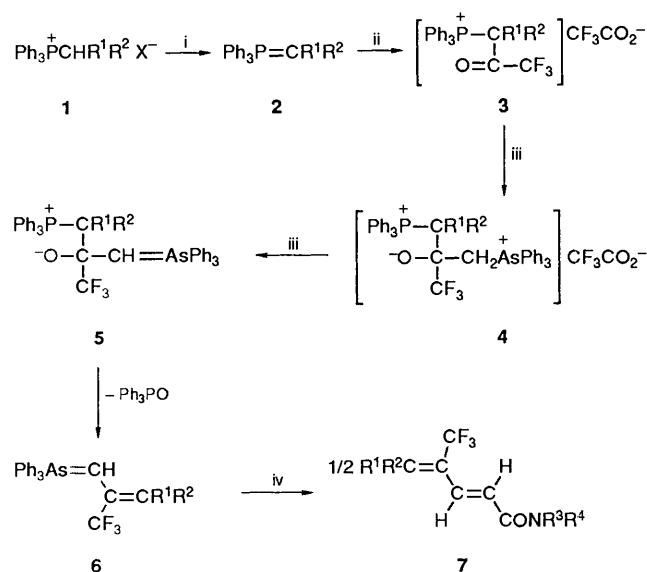
Shanghai Institute of Organic Chemistry, Academia Sinica, 345 Lingling Lu, Shanghai 200032, China

4-Trifluoromethyl-2,4-dienamides can be successfully synthesized by the reaction of trifluoromethylated arsoranes, generated from the transylidation between fluorinated β -oxoalkylphosphonium salts and methylenetriphenylarsorane, with α -bromoacetamides.

Synthesis of 2,4-dienamides is attracting much interest since such compounds are an important structural feature of a number of naturally occurring compounds which have been reported to be active both physiologically and insecticidally.¹ Several synthetic methods have been reported for their preparation.^{1,2} However, the methods for the preparation of trifluoromethyl analogues are still limited. The synthesis of 3-trifluoromethyl analogues from trifluoromethylated allylic alcohols and a (phenylthio)ynamine in several steps was reported by Kobayashi and co-workers.³ They would be expected to possess some biological activity and an effective method for their synthesis would be valuable.

Results and Discussion

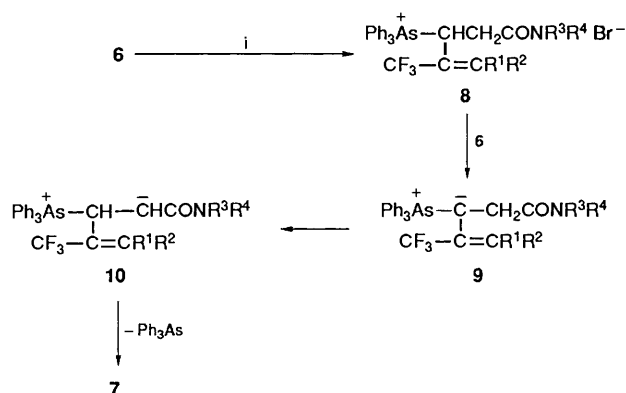
Recently we found a novel double elimination of arsonium salts and its application to the synthesis of 4-trifluoromethyl-2,4-dienyl carboxylates.⁴ In our continuing investigation into the exploitation of the synthetic utility of this reaction in organic synthesis we report a novel one-pot synthesis of 4-trifluoromethyl-2,4-dienamides by the reaction of trifluoromethylated arsoranes with α -bromoacetamides. The reaction sequence is shown in Scheme 1.



Scheme 1 Reagents: i, PhLi; ii, $(\text{CF}_3\text{CO})_2\text{O}$; iii, $\text{Ph}_3\text{As}=\text{CH}_2$; iv, $\frac{1}{2} \text{BrCH}_2\text{CONR}^3\text{R}^4$

The phosphoranes **2** generated from the corresponding phosphonium salts **1** and phenyllithium in tetrahydrofuran (THF) were acylated by the addition of trifluoroacetic anhydride (TFAA) to give the fluorinated β -oxophosphonium salts **3**, which in the reaction medium were attacked by methylenetriphenylarsorane to give intermediates **4**, followed by deprotonation (to species **5**) and elimination of triphenyl-

phosphine oxide to afford compounds **6**.⁵ Without isolation, compounds **6** reacted with α -bromoacetamides to give 4-trifluoromethyl-2,4-dienamides in 45–85% yield (3 steps). The mechanism of the reaction **6** \rightarrow **7** may be rationalized as in Scheme 2. The reaction is initiated by nucleophilic attack of



Scheme 2 Reagent: i, $\text{BrCH}_2\text{CONR}^3\text{R}^4$

trifluoromethylated arsoranes **6** on the α -carbon atom of the bromoacetamide to give arsonium salts **8**, which then react with another molecule of arsorane **6** to give the ylides **9**, which are in turn converted into the isomeric internal salts **10** via hydrogen transfer; this is then followed by elimination of triphenylarsine to afford products **7**. It may be that the internal salts **10** are more stable than the isomeric ylides **9** since the negative charge in species **10** can be stabilized by the CONR^3R^4 group.

Experimental

All b.p.s are uncorrected. IR spectra of liquid products were obtained as films on a Shimadzu IR-440 spectrometer. ^1H and ^{19}F NMR spectra were recorded on a Varian EM-360 (60 MHz) or XL-200 (200 MHz) spectrometer with SiMe_4 or $\text{CF}_3\text{CO}_2\text{H}$ (positive for upfield shifts) as external reference, respectively. J -Values are given in Hz. Mass spectra were measured on a GC-MS-4021 spectrometer.

General Procedure.—A solution of phenyllithium (3.0 mmol) in absolute diethyl ether (10 cm^3) was added dropwise to a stirred suspension of a phosphonium salt **1** (3.0 mmol) in absolute THF (30 cm^3) at -20°C under nitrogen. The mixture was stirred for 30 min at -20°C and was then cooled to -78°C ; TFAA (2.6 mmol) was slowly added until the characteristic colour of the phosphorane **2** had disappeared. After the mixture had been stirred at -78°C for 15 min, a solution of methylenetriphenylarsorane [generated from methyltriphenylarsonium iodide (6 mmol) and phenyllithium (6 mmol) in diethyl ether (30 cm^3) at 20°C for 1 h] was slowly added during 30 min. The mixture was then allowed to warm to 0°C , stirred for further 30 min, recooled to -78°C and the 2-bromoacetamide (1.3 mmol) was added. After the mixture had

Table 1 Preparation of 4-(trifluoromethyl)-2,4-dienamides

Compound	R ¹	R ²	R ³	R ⁴	Yield (%) ^a
7a	Me	Me	Et	Et	59
7b	Me	Me	Pr ⁱ	Pr ⁱ	45
7c	Me	Me	Bu	Bu	75
7d	Me	Me	-[CH ₂] ₅ -		57
7e	-[CH ₂] ₄ -		Et	Et	55
7f	-[CH ₂] ₄ -		Pr ⁱ	Pr ⁱ	51
7g	-[CH ₂] ₄ -		Bu	Bu	85
7h	-[CH ₂] ₄ -		-[CH ₂] ₅ -		51

^a Isolated yield. All products were characterized by microanalysis, and IR, NMR and mass spectroscopy.

been stirred at 20 °C for 2 h, the product **7** was isolated by column chromatography on silica gel with light petroleum (b.p. range 60–90 °C)–ethyl acetate (8:2) as eluent. The following dienamides were thus prepared (see Table 1 for definition of substituent groups).

N,N-Diethyl-5-methyl-4-(trifluoromethyl)hexa-2E,4-dienamide **7a** had b.p. 92 °C at 2 mmHg; δ_{H} 1.21 (6 H, t, *J* 6), 2.02 (3 H, s), 2.08 (3 H, s), 3.33 (4 H, q, *J* 6), 6.28 (1 H, d, *J* 16) and 7.36 (1 H, d, *J* 16); δ_{F} -20.7 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 250 ($\text{M}^+ + 1$, 35%), 249 (M^+ , 28), 234 ($\text{M}^+ - \text{Me}$, 31) and 177 ($\text{M}^+ - \text{NEt}_2$, 100) (Found: C, 58.2; H, 7.2; N, 5.3. C₁₂H₁₈F₃NO requires C, 57.85; H, 7.22; N, 5.62%).

N,N-Diisopropyl-5-methyl-4-(trifluoromethyl)hexa-2E,4-dienamide **7b** had b.p. 90 °C at 2 mmHg; δ_{H} 1.10 (12 H, d, *J* 6), 1.85 (3 H, s), 1.90 (3 H, s), 3.70 (2 H, septet, *J* 6), 6.16 (1 H, d, *J* 16) and 7.13 (1 H, d, *J* 16); δ_{F} -21.1 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 278 ($\text{M}^+ + 1$, 75%), 277 (M^+ , 11), 262 ($\text{M}^+ - \text{Me}$, 11) and 177 ($\text{M}^+ - \text{NPr}^i$, 97) (Found: C, 59.9; H, 7.85; N, 4.7. C₁₄H₂₂F₃NO requires C, 60.67; H, 7.94; N, 5.05%).

N,N-Dibutyl-5-methyl-4-(trifluoromethyl)hexa-2E,4-dienamide **7c** had b.p. 92 °C at 2 mmHg; δ_{H} 0.95 (6 H, t, *J* 6), 1.20–1.60 (8 H, m), 2.07 (3 H, s), 2.10 (3 H, s), 3.30 (4 H, t, *J* 6), 6.33 (1 H, d, *J* 16) and 7.41 (1 H, d, *J* 16); δ_{F} -21.0 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 306 ($\text{M}^+ + 1$, 48%), 305 (M^+ , 6), 262 ($\text{M}^+ - \text{Pr}$, 17) and 177 ($\text{M}^+ - \text{NBu}_2$, 100) (Found: C, 62.4; H, 8.7; N, 4.5. C₁₆H₂₆F₃NO requires C, 62.97; H, 8.52; N, 4.59%).

N-[5-Methyl-4-(trifluoromethyl)hexa-2E,4-dienoyl]piperidine **7d** had b.p. 106 °C at 2 mmHg; δ_{H} 1.50–1.70 (6 H, m), 1.99 (3 H, s), 2.04 (3 H, s), 3.50–3.64 (4 H, m), 6.16 (1 H, d, *J* 16) and 7.24 (1 H, d, *J* 16); δ_{F} -21.2 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 262 ($\text{M}^+ + 1$, 80%), 261 (M^+ , 26), 246 ($\text{M}^+ - \text{Me}$, 14), 177 ($\text{M}^+ - \text{NC}_5\text{H}_{10}$, 18) and 84 (NC_5H_{10} , 100) (Found: C, 59.4; H, 7.2, N, 5.4. C₁₃H₁₈F₃NO requires C, 59.79; H, 6.89; N, 5.36%).

4-Cyclopentylidene-*N,N*-diethyl-5,5,5-trifluoropenta-2E,4-dienamide **7e** had b.p. 102 °C at 2 mmHg; δ_{H} 1.05 (6 H, t, *J* 6), 1.50–1.70 (4 H, m), 2.23–2.73 (4 H, m), 3.25 (4 H, q, *J* 6), 6.16 (1 H, d, *J* 16) and 7.20 (1 H, d, *J* 16); δ_{F} -21.0 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 276 ($\text{M}^+ + 1$, 100%), 275 (M^+ , 36) and 203 ($\text{M}^+ - \text{NEt}_2$, 50) (Found: C, 61.3; H, 7.5; N, 5.1. C₁₄H₂₀F₃NO requires C, 61.11; H, 7.27; N, 5.09%).

4-Cyclopentylidene-5,5,5-trifluoro-*N,N*-diisopropylpenta-2E,4-dienamide **7f** had b.p. 96 °C at 2 mmHg; δ_{H} 1.30 (12 H, d, *J* 6), 1.62–2.00 (4 H, m), 2.55–2.82 (4 H, m), 3.93 (2 H, septet, *J* 6), 6.39 (1 H, d, *J* 16) and 7.32 (1 H, d, *J* 16); δ_{F} -20.3 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 304 ($\text{M}^+ + 1$, 45%), 303 (M^+ , 14) and 203 ($\text{M}^+ - \text{NPr}^i$, 60) (Found: C, 64.1; H, 8.0; N, 4.55. C₁₆H₂₄F₃NO requires C, 63.89; H, 7.92; N, 4.62%).

N,N-Dibutyl-4-cyclopentylidene-5,5,5-trifluoropenta-2E,4-dienamide **7g** had b.p. 102 °C at 2 mmHg; δ_{H} 0.95 (6 H, t, *J* 6), 1.20–1.94 (12 H, m), 2.54–2.87 (4 H, m), 3.00 (4 H, t, *J* 6), 6.37 (1 H, d, *J* 16) and 7.42 (1 H, d, *J* 16); δ_{F} -20.5 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 332 ($\text{M}^+ + 1$, 83%), 331 (M^+ , 11) and 203 ($\text{M}^+ - \text{NBu}_2$, 88) (Found: C, 64.9; H, 8.3; N, 4.2. C₁₈H₂₈F₃NO requires C, 65.28; H, 8.45; N, 4.23%).

N-(4-Cyclopentylidene-5,5,5-trifluoropenta-2E,4-dienoyl)-piperidine **7h** had b.p. 108 °C at 2 mmHg; δ_{H} 1.50–1.90 (10 H, m), 2.55–2.90 (4 H, m), 3.50–3.65 (4 H, m), 6.44 (1 H, d, *J* 16) and 7.38 (1 H, d, *J* 16); δ_{F} -18.7 (s); $\nu_{\text{max}}/\text{cm}^{-1}$ 1640s and 1600s; *m/z* 288 ($\text{M}^+ + 1$, 24%), 287 (M^+ , 26); 203 ($\text{M}^+ - \text{NC}_5\text{H}_{10}$, 6) and 84 (NC_5H_{10} , 100) (Found: C, 62.9; H, 7.1; N, 4.4. C₁₅H₂₀F₃NO requires C, 62.73; H, 6.96; N, 4.88%).

Acknowledgement

Thanks are due to the National Natural Science Foundation of China and Academia Sinica for financial support.

References

- 1 T. Mandai, T. Moriyama, K. Tsujimoto and J. Otera, *Tetrahedron Lett.*, 1986, **27**, 603 and references cited therein; T. Moriyama, T. Mandai, M. Kawada, J. Otera and B. M. Trost, *J. Org. Chem.*, 1986, **51**, 3896.
- 2 S. Tsuboi, Y. Nooda and A. Takeda, *J. Org. Chem.*, 1984, **49**, 1204; Y. Z. Huang, L. Shi, J. Yang and J. Zhang, *Tetrahedron Lett.*, 1987, **28**, 2195; D.-W. Ma and X.-Y. Lu, *Tetrahedron Lett.*, 1990, **46**, 3189.
- 3 Y. Hanzawa, K.-I. Kawagoe, A. Yamadam and Y. Kobayashi, *Tetrahedron Lett.*, 1985, **26**, 219.
- 4 Y.-C. Shen and Y.-J. Xiang, *Tetrahedron Lett.*, 1990, **31**, 2305.
- 5 Y.-C. Shen, Q.-M. Liao and W.-M. Qiu, *J. Chem. Soc., Chem. Commun.*, 1988, 1309.

Paper 1/01450A

Received 26th March 1991

Accepted 12th April 1991